

19.

Which set of compounds contains a salt that is soluble?

- ☒  $\text{CdCO}_3$ ,  $\text{PbSO}_4$ ,  $(\text{NH}_4)_2\text{CO}_3$
- ☐  $\text{CdCO}_3$ ,  $\text{PbSO}_4$ ,  $\text{ZnS}$
- ☐  $\text{Hg}_2\text{Br}_2$ ,  $\text{Ni}(\text{OH})_2$ ,  $\text{Cr}_2\text{S}_3$
- ☐  $\text{BaCrO}_4$ ,  $\text{AgCl}$ ,  $\text{Fe}(\text{OH})_3$
- ☐  $\text{Ca}_3(\text{PO}_4)_2$ ,  $\text{ZnO}$ ,  $\text{ZnCO}_3$

19.

INCORRECT

Which set of compounds contains a salt that is soluble?

**Your Answer:**  $\text{Ca}_3(\text{PO}_4)_2$ ,  $\text{ZnO}$ ,  $\text{ZnCO}_3$ **Correct Answer:**  $\text{CdCO}_3$ ,  $\text{PbSO}_4$ ,  $(\text{NH}_4)_2\text{CO}_3$ 

All of these salts are insoluble.